### organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.058 wR factor = 0.105 Data-to-parameter ratio = 15.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## 2-(Hydroxymethyl)-1,2-benzothiazol-3(2H)-one

In the title compound,  $C_8H_7NO_2S$ , the benzisothiazolone ring system is planar, with a maximum deviation from the mean plane of 0.038 (2) Å for the N atom. Weak intermolecular hydrogen bonds and a  $\pi$ - $\pi$  stacking interaction stabilize the crystal structure.

#### Comment

It has been reported that isothiazolones and heterocyclic bioisosteric derivatives are potent industrial microbiocides with antifungal and antibacterial activities (Taubert *et al.*, 2002). These compounds are also highly potent platelet aggregation inhibitors (Vicini *et al.*, 2000), and show antiinflammatory, analgesic and antipyretic activities (Bordi *et al.*, 1992). In view of this, we report here the crystal structure of the title compound, (I) (Fig. 1).



Bond lengths and angles of the benzisothiazolone ring system (Table 1) are in good agreement with the values quoted in a previous report (Kim *et al.*, 1996). The benzisothiazolone ring system is planar, with a maximum deviation from the mean plane of 0.038 (2) Å for atom N1.

The crystal structure of (I) is stabilized by weak intermolecular hydrogen bonds (Table 2) and  $\pi$ - $\pi$  stacking interactions. The isothiazolone rings of centrosymmetrically





**Figure 1** View of the title compound, with displacement ellipsoids drawn at the 40% probability level. Received 9 December 2005 Accepted 21 December 2005 Online 7 January 2006





A packing diagram of the molecule of the title compound, viewed approximately down the b axis. Hydrogen bonds are shown as dashed lines.

related molecules at (x, y, z) and (-x, -y, -z) stack with a distance of 3.496 (3) Å between the ring centroids.

### **Experimental**

A mixture of benzisothiazolone (0.02 mol) and polyoxymethylene (0.022 mol) was stirred in refluxing chloroform (10 ml) for 5 h at 334 K to afford the title compound (3.13 g, yield 87%). Single crystals suitable for X-ray measurements were obtained by recrystallization from ethanol at room temperature.

Crystal data

$C_8H_7NO_2S$	Mo $K\alpha$ radiation
$M_r = 181.21$	Cell parameters from 25
Tetragonal, $I4_1/a$	reflections
a = 17.856 (7)  Å	$\theta = 4 - 14^{\circ}$
c = 9.955 (2) Å	$\mu = 0.36 \text{ mm}^{-1}$
$V = 3174.0 (19) \text{ Å}^3$	T = 293 (2) K
Z = 16	Block, colourless
$D_x = 1.517 \text{ Mg m}^{-3}$	$0.25 \times 0.20 \times 0.18 \ \text{mm}$
Data collection	
Enraf-Nonius CAD-4	$R_{\rm int} = 0.076$
diffractometer	$\theta_{\rm max} = 27.0^{\circ}$
$\omega$ scans	$h = -22 \rightarrow 22$
Absorption correction: $\psi$ scan	$k = -22 \rightarrow 22$
(North et al., 1968)	$l = 0 \rightarrow 11$
$T_{\min} = 0.915, T_{\max} = 0.942$	3 standard reflections
6764 measured reflections	every 100 reflections
1704 independent reflections	intensity decay: none
1365 reflections with $I > 2\sigma(I)$	
Refinement	
Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0408P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.058$	+ 0.6111P]
$wR(F^2) = 0.105$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.12	$(\Delta/\sigma)_{\rm max} < 0.001$
1704 reflections	$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
110 parameters	$\Delta \rho_{\rm min} = -0.99 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHEL

d Extinction correction: *SHELXL97* Extinction coefficient: 0.0021 (5)

Table 1				
Selected	geometric parameters	(Å,	°)	).

8	I		
S1-N1	1.7119 (17)	N1-C7	1.372 (2)
S1-C1	1.7335 (18)	N1-C8	1.456 (2)
O1-C8	1.391 (2)	C7-O2	1.234 (2)
O2-C7	1.234 (2)		
N1-S1-C1	90.51 (8)	C8-N1-S1	119.23 (13)
C7-N1-C8	123.83 (16)	O1-C8-N1	112.77 (15)
C7-N1-S1	116.09 (11)		. ,
C1-S1-N1-C7	2.77 (13)	N1-S1-C1-C2	179.03 (16)
C1-S1-N1-C8	172.64 (14)	C7-N1-C8-O1	94.6 (2)
N1-S1-C1-C6	-2.26 (13)	S1-N1-C8-O1	-74.5 (2)

Table 2		
Hydrogen-bond geometry	(Å.	°)

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O1-H1A\cdots O2^{i}$	0.82	1.97	2.774 (2)	169
$C5-H5A\cdots O2^{n}$	0.93	2.57	3.492 (3)	171

Symmetry codes: (i)  $y + \frac{1}{4}$ ,  $-x + \frac{1}{4}$ ,  $z + \frac{1}{4}$ ; (ii)  $-y + \frac{1}{4}$ ,  $x - \frac{1}{4}$ ,  $z - \frac{1}{4}$ .

All H atoms were placed in calculated positions, with C–H = 0.93–0.97 Å and O–H = 0.82 Å, and included in the final cycles of refinement using a riding model, with  $U_{\rm iso}(\rm H) = 1.2 U_{eq}(\rm C)$  for the aryl and methylene H atoms and  $1.5 U_{eq}(\rm O)$  for the hydroxy H atom.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *NRCVAX* (Gabe *et al.*, 1989); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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